

## REVIEW

# Phenolic compounds of the genus *Iris* plants (*Iridaceae*)

## Fenolické sloučeniny rostlin rodu *Iris* (*Iridaceae*)

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### Summary

This article presents the results of testing of phenolic compounds (flavonoids, isoflavonoids, xanthenes, phenolcarboxylic acids, tannins, coumarins, etc.) in the rhizomes of four *Iris* species (*Iris sibirica* L., *Iris pseudacorus* L., *Iris imbricat* Lindl., *Iris hungarica* Waldst. et Kit.). With the use of paper and thin-layer chromatography, fifteen phenolic compounds were identified: gallic, coumaric, cinnamic, chlorogenic, neochlorogenic, ferulic, caffeic acids; kaempferol, quercetin, hispidulin, daidzein, genistein, formononetin, mangiferin and isomangiferin. Quantitative contents of flavonoids (1.2–3.7%), hydroxycinnamic acids (0.6–6.5%),  $\gamma$ -pyrones (0.01–0.8%), tannins (6–14%), isoflavonoids (1–2%), polyphenolic compounds (up to 3%) in the rhizomes of the *Iris* species were determined. Chosen plants belong to natural flora and have been often cultivated. However, this phytochemical analysis for the main groups of the biologically active substances shows a perspective use of the *Iris* species in medicine.

**Key words:** *Iris* species • *Iridaceae* • phenolic compounds • chromatography • qualitative analysis • quantitative contents

### Souhrn

Článek shrnuje výsledky testování fenolických sloučenin (flavonoidy, isoflavonoidy, xanthy, fenolkarboxylové kyseliny, taniny, kumariny atd.) obsažených v oddencích čtyř druhů kosatců (*Iris sibirica* L., *Iris pseudacorus* L., *Iris imbricat* Lindl., *Iris hungarica* Waldst. et Kit.).

S využitím papírové a tenkovrstvé chromatografie bylo identifikováno patnáct fenolických sloučenin: kyselina gallová, kumarová, skořicová, chlorogenová, neochlorogenová, ferulová, kávová; kaempferol, quercetin, hispidulin, daidzein, genistein, formononetin, mangiferin a isomangiferin. Byly stanoveny kvantitativní obsahy flavonoidů (1,2–3,7 %), hydroxyskořicových kyselin (0,6–6,5 %),  $\gamma$ -pyronů (0,01–0,8 %), taninů (6–14 %), isoflavonoidů (1–2 %) a polyfenolických sloučenin (až 3 %) v oddencích druhů kosatců. Vybrané rostliny patří k přírodní flóře a jsou často pěstovány. Nicméně tato fytochemická analýza pro hlavní skupiny biologicky aktivních látek ukazuje na perspektivní využití druhů *Iris* v lékařství.

**Klíčová slova:** druhy *Iris* • *Iridaceae* • fenolické sloučeniny • chromatografie • kvalitativní analýza • kvantitativní obsah

### Introduction

Phenolic compounds are the aromatic compounds that have one or more bound hydroxyl groups as a substituent in the aromatic ring; this is one of the most widely-distributed classes of natural compounds exhibiting biological activity. Phenolic compounds are formed almost in every plant cell and participate in the basic life processes of plants: respiration, photosynthesis, the formation of cell walls, protection of the plant from diseases, viruses, the action of abiotic and biotic stresses<sup>1</sup>. Phenolic and polyphenolic compounds provide antioxidant, anti-inflammatory, capillary-restorative, anti-cancerous, hepatoprotective effect<sup>2</sup>. The most widely-distributed in nature are caffeic acid and its derivatives (chlorogenic acid and its isomers) that provide anti-inflammatory and cholagogue effect. Ferulic acid is a part of the plant cells. Chlorogenic, ferulic, caffeic, coumaric acids together provide gipoazotemic effect, increase renal function, stimulate antitoxic function of the liver<sup>3, 4</sup>. Coumarins are used as anticoagulants, providing bactericidal, fungicidal, anti-cancerous activity<sup>5, 6</sup>. Isoflavonoids have antioxidant, antibacterial, anti-inflammatory and estrogenic effects<sup>4, 7, 8</sup>.

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The objects of the study were *Iris* (*Iridaceae*)<sup>9–11</sup>. *Iris* accumulate various secondary metabolites: flavonoids<sup>12</sup>, isoflavonoids and their glycosides, benzoquinones, triterpenoids, stilbens<sup>13–15</sup>; they provide anti-inflammatory, antioxidant, antituberculosis, diuretic and other effects<sup>2, 16, 17</sup>. The genus *Iris* plants grows on the territory of Ukraine<sup>18</sup> and contiguous countries<sup>19</sup> in large quantities. The richest diversity of wild *Iris* species is not only the basis for the introduction of valuable species into the culture, but also a source of biologically active substances (BAS). The aim of the study is to perform qualitative and quantitative analysis of the phenolic compounds in the rhizomes of the four *Iris* species.

## Experimental part

### Plant material

In the preliminary investigation, twenty-one samples of materials (leaves, rhizomes) were analysed, including eleven *Iris* species and four *Iris* varieties. Species and varieties were harvested from the collections of the following botanical gardens: V. N. Karazin Kharkiv National University (Kharkiv, Ukraine; 2012, 2014), M. M. Gryshko National Botanical Garden of the National Academy of Sciences of Ukraine (Kiev, Ukraine; 2012, 2014, 2015), National University of Pharmacy (Kharkiv, Ukraine, 2007–2009), Azerbaijan Medical University (Baku, Azerbaijan, 2005) and in the surrounding area of Kharkiv.

The raw material was dried to air-dry state, cut into pieces, packed in cloth bags and stored in a cool, dry place. For chemical analysis an average sample of raw material was reduced to the particle size of 2–3 mm<sup>20–23</sup>.

### Qualitative analysis

Qualitative analysis of the phenolic compounds of the studied species of irises was performed using generally accepted methods and techniques of phytochemical analysis. For preliminary analysis, the rhizomes of irises were exposed to exhaustive extraction of the raw materials with water, 20% and 70% ethanol (in the ratio of the raw material to the extractant in the proportion of 1 : 5) in a boiling water bath with the reverse refrigerator. The obtained extractions were concentrated by evaporation of the solvent in a water bath (to decrease volume by half). Further it was used for qualitative reactions and chromatographic analysis<sup>1, 4–6, 8</sup>.

Coumarins were determined according to the results of the reactions: lactone reaction (yellow colouring) and formation of azo dye (red colouring).

Glycosides and aglycons of flavonoids were determined in hydro-alcoholic extracts in the reactions of identification: cyanidine reaction by Bryant (yellow-red colouring of the aqueous phase and yellow-hot colouring of the octal phase), the reaction with 3% solution of iron (III) chloride (dark green colour of flavonols, flavones); the reaction with an alkaline solution (bright yellow colour); the reaction with 5% solution of aluminium chloride (yellow-green colouring); the boric-acid reaction (yellow colouring on detection of 3- and 5-hydroxyflavones and 5-hydroxyflavanones); the reaction with ammonia (flavones, flavonols, flavanones and

flavanonols dissolve with formation of yellow colour, which, when heated, changes to orange or brown colour).

To determine tannins, the reactions of the sediment were carried out with 1% gelatin solution, 1% solution of quinine hydrochloride, 10% solution of basic acetate of lead. The group of tannins was detected by the reaction with a solution of iron ammonium alum.

Chromatographic tests were carried out on the paper «Filtrak» (FN-1, FN-4), for thin-layer chromatography using silica gel 60 F<sub>254</sub> TLC plates (by “Mersk”, Germany). The process of chromatography was carried out by the methods of ascending, descending unidimensional, two-dimensional chromatography. The solvents relation is indicated by figures and taken in volumetric units.

The systems of solvents were used for chromatography:

- 1 – *n*-butanol – acetic acid – water (4 : 1 : 2 by volume) is for testing of the phenolic compounds, flavonoids, phenolcarboxylic acids, xanthenes
- 2 – 15% acetic acid is for determining phenolcarboxylic acids, xanthenes
- 3 – 2% acetic acid is for determining phenolcarboxylic acids
- 4 – chloroform – acetic acid – water (13 : 6 : 1 by volume) is for determining flavonoids, isoflavonoids
- 5 – benzole – ethyl acetate – acetic acid – water (50 : 50 : 1 : 1 by volume) is similar No 4
- 6 – chloroform (formamide 25%) is for determining coumarins
- 7 – petroleum ether (formamide 25%) is for determining coumarins
- 8 – benzole (formamide 25%) is for determining coumarins
- 9 – benzole – ethyl acetate (2 : 1 by volume) – is for determining coumarins; TLC
- 10 – ethyl acetate – formic acid – glacial acetic acid – water (100 : 11 : 11 : 26 by volume) – is for determining flavonoids, xanthenes; TLC
- 11 – chloroform – methanol (8 : 2 by volume) is for determining flavonoids, isoflavonoids, phenolcarboxylic acids; TLC

On the chromatograms the substances were detected by the characteristic fluorescence in UV-light at the wavelength of 365 nm and 254 nm before and after processing the chromatograms with ammonium vapours, 2% alcoholic solution of aluminum chloride; 10% sodium/potassium hydroxide solution; 5% alcoholic solution of diazotized sulfanilic acid (diazo reagent) (for determining coumarins, hydroxycinnamic acids) that allows to obtain bands with the brighter fluorescence in UV-light. For the identification of the substances, the value of  $R_f$  was determined and compared with the standard samples (flavonoids aglycones and isoflavonoids, xanthenes, phenolcarboxylic acids).

For further analysis of the flavonoids aglycones and isoflavonoids, acid hydrolysis of the sum of the phenolic compounds was conducted with 5% sulfuric acid for 5 hours in a water bath. After cooling, the solution was transferred into a separatory funnel and worked with ethyl acetate (6 times and 10 ml by volume). The ethyl acetate extractions were combined, then transferred to the

separatory funnel and processed with water to eliminate residua of acid (by the universal indicator). Received ethyl acetate extractions were evaporated and 2 ml of 70% alcohol were added. Analysis of the substances was carried out by PC and TLC systems: 1, 4, 5, 10, 11.

#### Extraction of raw material

Air-dried rhizomes of *Iris pseudacorus* (1.0 kg) were extracted with EtOH (50%) in a percolator for 24 h. The extraction was repeated twice under the same conditions. The aqueous EtOH extracts were combined, filtered, evaporated in a rotary evaporator to 0.7 L of aqueous residue, and left for 1 d. The supernatant liquid was separated. The resulting extract was worked up successively with  $\text{CHCl}_3$ , EtOAc, and *n*-BuOH. The resulting extracts were evaporated *in vacuo*. The qualitative composition of chloroform, EtOAc and butanol fractions controlled by paper chromatography in a solvent system number 1, 2, 3. In aqueous residue, substances **1–4** were identified by preparative chromatography. EtOAc and butanol extracts were combined and subjected to CC on silica gel and eluted with gradient:  $\text{CHCl}_3$  and ethanol-mixtures with increasing ethanol concentration (0–100%) to afford 152 fractions. After analysis, the presence of compounds **5–10** was found.

#### Experimental

When determining the structures of received substances, physical and physicochemical methods of analysis (UV-, IR-,  $^1\text{H-NMR}$  spectroscopy, mass spectrometry, chromatography in thin layer of sorbent (Silufol UV-254 plates; silica gel 60  $F_{254}$  TLC plates), column chromatography on silica gel of Silica Gel 100–200 (75–150 mmm) (US)) were used. Silica gel was pre-purified from ions of metals with 0.5% solution of chloric acid, then it was dried at room temperature and then it was activated at a temperature of 130–140 °C for 2 hours in a drying cabinet. Melting point (MP) was determined on the Kofler block (Franz Kustner nqch K:G:Dresden; N.K.70/3314k). The recrystallization of selected substances was carried out in 96% ethanol with an addition of 2–3 drops of water. The substances were dried under vacuum (10–2 mm Hg) over  $\text{P}_2\text{O}_5$  at 110 až 115 °C for 5 hours.

UV-absorption spectra and optical density were recorded with a spectrophotometer module of Carl Zeiss (Germany) in cells with layer thickness of 10 mm. IR-spectra were recorded on the instrument Tensor 27, UR-20 (GDR) in the tablets of potassium bromide at the ratio of substance and filler of 1 : 200 – 1 : 400.  $^1\text{H-NMR}$ -spectra were recorded with the device Varian Mercury-VX-200 (200 MHz<sub>2</sub>) (USA), the solvent  $\text{DMSO-}d_6$  (the internal standard TMS). Mass-spectra were measured on the instrument Varian 1200 L (USA) (the temperature of the ionization chamber is 150–300 °C, the ionizing voltage is 70 eV, 40–600 m/z).

#### Compounds

**Caffeic acids 1.**  $\text{C}_9\text{H}_8\text{O}_4$ , mp 194–195 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 325, 300, 235.  $R_f$  0.32, system 3;  $R_f$  0.92, system 11. IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 3400, 3235, 2975 (OH),

1647, 1630 (C = O), 1607, 1540 (Ar), 880, 860 (substituted benzene).

**Ferulic acids 2.**  $\text{C}_{10}\text{H}_{10}\text{O}_4$ , mp 168–170 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 320, 290sh.  $R_f$  0.43, system 3;  $R_f$  0.18, system 11.

**Chlorogenic acids 3.**  $\text{C}_{16}\text{H}_{18}\text{O}_9$ , mp 203–205 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 325, 300, 245.  $R_f$  0.66, system 3; IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 2970, 2900, 2780 (OH), 1740–1710 (ester group), 1660, 1625 (C = O), 1605–1550 (Ar), 840 (substituted benzene).

**Neochlorogenic acids 4.**  $\text{C}_{16}\text{H}_{18}\text{O}_9$ , amorphous solid; UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 325, 300, 243.  $R_f$  0.70, system 3.

**Kaempferol 5.**  $\text{C}_{15}\text{H}_{10}\text{O}_6$ , mp 275–277 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 265, 369.  $R_f$  0.55, system 4;  $R_f$  0.63, system 5. IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 3340 (-OH), 1660 (C = O), 1615, 1570, 1510 (C = C), 820 (*n*-substitution in the ring “B”).  $^1\text{H NMR}$  (200 MHz,  $\text{DMSO-}d_6$ , TMS):  $\delta$  6.11 (1H, s, 6-H),  $\delta$  4 (1H s, 8-H), 7.90 (2H, d, H 2', 6'), 6.80 (2H, d, H-3', 5').

**Hispidulin 6.**  $\text{C}_{16}\text{H}_{12}\text{O}_6$ , mp 287–290 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 336, 275.  $R_f$  0.86, system 5;  $R_f$  0.39, system 10. IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 3300, 3100 (-OH), 1655 (C = O), 1610, 1570, 1490 (C = C), 2950–2850 (-OCH<sub>3</sub>).  $^1\text{H NMR}$  (200 MHz,  $\text{DMSO-}d_6$ , TMS):  $\delta$  6.6 (1H, s, 3-H), 13.1 (1H, s, 5-H), 6.7 (1H, s, 8-H), 7.9 (2H, d, H 2', 6'), 6.9 (2H, d, H-3', 5').

**Genistein 7.**  $\text{C}_{15}\text{H}_{10}\text{O}_5$ , mp 290–292 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 260, 325.  $R_f$  0.88, system 5;  $R_f$  0.41, system 11. IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 3726, 3410, 3104 (-OH), 1651 (C = O), 1615, 1569, 1424, 1503 (C = C), 885 (*n*-substitution in the ring “B”).  $^1\text{H NMR}$  (200 MHz,  $\text{DMSO-}d_6$ , TMS):  $\delta$  8.42 (s, 2-H), 6.72 (d, 6-H), 6.47 (d, 8-H), 7.40 (d, H 2', 6'), 6.83 (d, H-3', 5'), 9.64 (s, OH-4'), 12.94 (s, OH-5).

**Daidzein 8.**  $\text{C}_{15}\text{H}_{10}\text{O}_4$ , mp 307–308 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 249 sh, 303.  $R_f$  0.77, system 4. IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 3666, 3171 (-OH), 1631 (C = O), 1595, 1518, 1188, 1460 (C = C), 887, 820 (*n*-substitution in the ring “B”).  $^1\text{H NMR}$  (200 MHz,  $\text{DMSO-}d_6$ , TMS):  $\delta$  8.31 (s, 2-H), 8.05 (d, 5-H), 7.14 (dd, 6-H), 7.23 (d, 8-H), 7.41 (d, H-2', 6'), 6.82 (m, H-3', 5').

**Mangiferin 9.**  $\text{C}_{19}\text{H}_{18}\text{O}_{11}$ , mp 267–269 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 364, 315, 257, 240.  $R_f$  0.57, system 1.  $R_f$  0.49, system 10. IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 1650 (C = O), 3100–3700 (-OH), 1591, 1565, 1494 (C = C).  $^1\text{H NMR}$  (200 MHz,  $\text{DMSO-}d_6$ , TMS):  $\delta$  6.35 (1H, s, 4-H), 6.83 (1H, s, 5-H), 7.38 (1H, s, 8-H), 13.77, 10.65, 9.85 (1H, s, OH).

**Isomangiferin 10.**  $\text{C}_{19}\text{H}_{18}\text{O}_{11}$ , mp 248–250 °C. UV  $\lambda_{\text{max}}$  ( $\text{C}_2\text{H}_5\text{OH}$ ) nm: 366, 314, 257, 240.  $R_f$  0.55, system 1.  $R_f$  0.44, system 10. IR (KBr),  $\nu$ ,  $\text{cm}^{-1}$ : 1650 (C = O), 3100–3700 (OH), 1591, 1565, 1494 (C = C);  $^1\text{H NMR}$  (200 MHz,  $\text{DMSO-}d_6$ , TMS):  $\delta$  6.88 (1H, s, 5-H), 7.45 (1H, s, 8-H), 6.25 (1H, s, 2-H), 13.78, 10.60, 9.85 (1H, s, OH).

#### Quantitative analysis

For the quantitative determination of the contents in the raw materials of flavonoids, phenolic compounds, hydroxycinnamic acids, isoflavonoids the spectrophotometric methods were used; for tannins, the titrimetric method was used in accordance with the

recommendations of the SP XI, SFU. For the quantitative determination of flavonoids, the method based on the reaction of complex formation of flavonoids with the solution of 3% aluminium chloride was used<sup>24</sup>, in terms of the standard pharmacopoeia sample (PSS) rutin (PA 42-2508-87; «Herb of Hypericum») at the wavelength of 410 nm<sup>23</sup>.

The content of the sum of the hydroxycinnamic acids was determined on conversion to the PSS chlorogenic acid at the wavelength of 327 nm, as the basis took the technique of “Herb of Erigeron canadensis” (TPA 42-U-6/37-323-96)<sup>23</sup>.

The content of the sum of the phenolic compounds was determined at the wavelength of 270 nm on conversion to the PSS gallic acid according to the method of standardization of the Piphamin drug on the basis of the herb of pea field<sup>25</sup>. For the analysis of tannins, the Neubauer-Löwenthal titrimetric method was used, which based on the oxidation of polyphenols with potassium permanganate<sup>20</sup>.

To establish the content of isoflavonoids, the method of quantitative determination of isoflavonoids in the roots of restharrow of the PA “Roots of restharrow” was taken as the basis<sup>25</sup>, but as the extractant 50% alcohol was used. The optical density of the extract was determined at the wavelength of 271 nm; the conversion of the content of isoflavonoids was conducted to onosid (PA-42U “Herb of haricot”)<sup>8</sup>, as the maximum rhizomes of irises solution absorption is on this wave length.

Quantitative determination of the sum of  $\gamma$ -pyrones in terms of mangiferin at the wavelength of 369 nm conducted with the previously published methods<sup>26</sup>.

UV-absorption spectra and optical density were recorded with a spectrophotometer module of SP-46, Carl Zess (Germany) Specord M-80, Thermo Scientific Evolution 60S UV – Visible Spectrophotometer (USA), in the wavelength range of 200–500 nm.

Statistical analysis of the results was conducted in accordance with the requirements of the State Pharmacopoeia of Ukraine, the 1<sup>st</sup> issue, supplement 1, p. 5.3 using Microsoft Office Excel 7.0 application.

## Results and discussion

### Primary screening

The presence of flavonoids, coumarins, hydroxycinnamic acids and tannins was established with the qualitative reactions and chromatographic analysis in the aqueous and hydro-alcoholic extracts of 21 samples of the leaves and rhizomes of irises (Table 1).

For the next study, four *Iris* species (*Iris hungarica* Waldst. et Kit., *Iris sibirica* L., *Iris imbricata* Lindl., *Iris pseudacorus* L.) (Fig. 1) that showed good results were chosen, they have a sufficient resource base and belong to different sections. According to the classification of Rodionenko (1961)<sup>27</sup>, *I. hungarica*, *I. imbricata* belong to the group of bearded irises (*Barbatae* are the species with flowers, bearing on the outer perianth lobes beard of

Table 1. Qualitative analysis of BAS in the leaves and rhizomes of Irises

Species/varieties of iris	Collecting year	Plant part	XAN	ISOF	FLA	COU	HYD.ac.	TAN
<i>Iris hungarica</i> Waldst. et Kit.	2014	rh	+++	++	+++	++	+++	+++
<i>Iris hungarica</i> Waldst. et Kit.	2014	l	+++	+++	+++	+	+++	++
<i>Iris sibirica</i> L.	2012	rh	+	++	+	+	++	+++
<i>Iris sibirica</i> L.	2012	l	+	+	++	+	++	++
<i>Iris pseudacorus</i> L.	2012	rh	+++	+++	+++	+++	++	+++
<i>Iris pseudacorus</i> L.	2012	l	+++	++	++	+	++	+
<i>Iris pseudacorus</i> f. <i>alba</i>	2012	rh	+	+	++	+	+	++
<i>Iris germanica</i> L.	2007	rh	++	+++	+++	++	+	++
<i>Iris germanica</i> L.	2007	l	++	++	+++	++	++	++
<i>Iris florentina</i> L.	2007	rh	++	++	+++	++	++	++
<i>Iris florentina</i> L.	2007	l	++	++	+++ + ++	++		
<i>Iris lactea</i> Pall	2015	rh	++	++	+++	++	++	++
<i>Iris lactea</i> Pall.	2015	l	++	+	++	+	++	+
<i>Iris Medwedewii</i> Fomin	2014	rh	+++	+	+	+	+	+
<i>Iris arthalinae</i> Fomin	2014	rh	++	+	+	+	+	+
<i>Iris imbricat</i> Lindl.	2005	rh	+++	++	+++	+	+++	++
<i>Iris pallida</i> Lam.	2012	rh	++	–	+	+	+	+
<i>Iris sibirica</i> “Snowcrest”	2012	rh	+	–	+	–	–	+
<i>Iris sibirica</i> “Gatinea”	2012	rh	+	+	+	–	–	–
<i>Iris sibirica</i> “Cambridge Blue”	2012	rh	+	–	+	–	–	–
<i>Iris sibirica</i> “Violet Fantasy”	2012	rh	+	–	+	–	–	–

rh – rhizomes, l – leaves, XAN – xanthonenes, ISOF – isoflavonoids, FLA – flavonoids, COU – coumarins, HYD.ac. – hydroxycinnamic acids, TAN – tannins

evaluation of reactions: – any reaction, + weak reaction, ++ strong reaction, but with some deficiencies in coloration, +++ best reaction

Fig. 1. General view of the living plants and studied raw material of the *Iris* species*Iris pseudacorus**Iris hungarica**Iris sibirica**Iris imbricata*

multicellular hairsprings) from section *Iris*, the series *Elatae* Lawrence. *I. pseudacorus* (series *Laevigatae* (Diels) Lawrence), *I. sibirica* (series *Sibiricae* (Diels) Lawrence) belong to the *Limniris* (Tausch) Spach em. Rodion section, to the group of beardless irises (*Imberbis* are the species with flowers, deprived of their beards).

The identification of phenolic compounds was carried out by paper chromatography to confirm the results of the qualitative reactions.

#### Identification of *Iris* phenolic compounds

Phenolic compounds were determined in the extracts of the rhizomes of *irises* by two-dimensional PC which is the most efficient for separation of phytochemical compounds.

The results of the chromatographic alcoholic extracts of *irises* analysis showed that when identified in UV-light and according to the results of the qualitative reactions with the alcoholic solutions of sodium hydroxide, aluminum chloride and diazoreagent on the chromatograms of *I. pseudacorus*, *I. imbricata*, *I. sibirica*, *I. hungarica* not less 8–21 substances of phenolic nature were found.

Some bands had yellow and light-yellow colour in UV-light that is specific to flavones and their 7-glycosides, aglycones. Bands with yellow-brown or dark-brown colour were characterized as flavones, flavone-3-glycosides, and flavanones. Bands with bright-orange fluorescence are specific to the derivatives of dibenzo- $\gamma$ -pyrones. Absorption bands with blue, violet, violet-blue fluorescence are specific to coumarins, phenolcarboxylic acids, isoflavonoids, flavone-5-glycosides, stilbens. Bands with red fluorescence are probably chalcones, aurones.

Chromatograms differed in the number of spots, their chromatographic behaviour, and colour of spots before and after they were exposed to UV-light,  $R_f$  values.

After processing chromatograms with ammonia vapours and 2% alcoholic solution of aluminum chloride spots of aglycones got bright-yellow fluorescence, dark-

brown spots got yellow-green colour that is specific of flavones glycosides. Acid hydrolysis was carried out to establish their nature.

**Flavonoids (C6 – C3 – C6).** Extracts were chromatographed in the systems 1, 4, 5, 10, 11. By the characteristic fluorescence in UV-light,  $R_f$  value and colour of spots on chromatograms after they were exposed to ammonia vapours and solution of aluminum chloride in comparison with authentic samples and literature data flavonoids aglycones and isoflavonoids were identified as quercetin, kaempferol, hispidulin, daidzein, formononetin, genistein (Table 2).

On the grounds of spots colour of isoflavonoids in UV-light coincided with some phenolic acids, further chromatographic analysis was carried out with authentic samples of substances, their colour and  $R_f$  values of the spots were compared.

**Phenolcarboxylic acids (6 – 1, 6 – 3).** Extracts from rhizomes of *irises* were chromatographed in the systems 1, 2, 3, 11. In UV-light at the wavelength of 365 nm after processing chromatograms with ammonia vapours spots became visible, which fluorescence intensified from blue to blue-green, after processing with diazoreagent solution the spots became reddish-brown, which is specific of hydroxycinnamic acids. When processing chromatograms with solution of aluminum chloride colour of spots intensified or changed.

By chromatographic behaviour (colour of spots in UV-light, colour after processing with chromogenic reagents,  $R_f$  values when compared with previously known substances) caffeic, cinnamic, *n*-coumaric, ferulic, chlorogenic, neochlorogenic, gallic acids were identified (Table 3).

**Coumarins (C6 – C3).** Extracts were chromatographed in the systems 6, 7, 8, 9. After processing chromatograms with 10% solution of potassium hydroxide and diazoreagent spots with purple, bluish-green, grey-green, green, yellow fluorescence were identified which indicated the presence of derivatives benzo- $\alpha$ -pyrones<sup>5</sup>.

Table 2. Chromatographic characterization of phenolcarboxylic acids identified in the rhizomes of *Iris*

Phenol carboxylic acids	R <sub>f</sub> value in systems			Glow in UV light				Species of iris, where the substance was found
	1	2	3	Before exposing	After NH <sub>4</sub> OH processing	After 2% AlCl <sub>3</sub> processing	After diazo reagent processing	
<i>n</i> -coumaric	0.74	0.60	0.48	blue	violet-blue	brown	cherry red	<i>I. hungarica</i> , <i>I. sibirica</i> , <i>I. imbricat</i>
Caffeic	0.80	0.48	0.32	light blue	bright blue	gray-green	red-brown	<i>I. pseudacorus</i> , <i>I. sibirica</i>
Ferulic	0.87	0.56	0.43	violet-blue	violet	brown	red-violet	<i>I. pseudacorus</i> , <i>I. imbricat</i> , <i>I. sibirica</i>
Chlorogenic	0.62	0.69	0.66	blue	green-blue	green	red-brown	<i>I. pseudacorus</i> , <i>I. imbricat</i> , <i>I. hungarica</i>
Neochlorogenic	0.64	0.75	0.70	blue	green-blue	green	dark red	<i>I. pseudacorus</i> , <i>I. imbricat</i>
Cinnamic	–	–	0.54	blue	violet-blue	brown	brown	<i>I. hungarica</i>
Gallic	0.65	0.60	–	light blue	dark blue	–	dark blue	<i>I. hungarica</i>

Table 3. Chromatographic characterization of phenolic substances identified in the rhizomes of *Iris*

Substances	R <sub>f</sub> value in systems			Glow in UV light			Species of irises, where the substance was found
	1	4	5	Before exposing	After NH <sub>4</sub> OH processing	After 2% AlCl <sub>3</sub> processing	
Kaempferol	0.86	0.55	0.63	pale yellow	yellow-green	brown	<i>I. pseudacorus</i> , <i>I. imbricata</i>
Hispidulin	–	–	0.86	dark brown	dark brown	yellow	<i>I. pseudacorus</i>
Quercetin	0.73	0.32	0.26	yellow	yellow-brown	yellow-green	<i>I. sibirica</i>
Daidzein	0.88	0.77	–	blue	bright blue	–	<i>I. imbricat</i> , <i>I. sibirica</i> , <i>I. hungarica</i> , <i>I. pseudacorus</i>
Genistein	0.70	–	0.88	dark	dark brown	yellow	<i>I. pseudacorus</i> , <i>I. hungarica</i>
Formononetin	0.94	–	–	blue	bright blue	–	<i>I. sibirica</i> , <i>I. imbricata</i> , <i>I. hungarica</i>
Mangiferin	0.57	–	–	dark yellow	bright orange	green	<i>I. imbricat</i> , <i>I. sibirica</i> , <i>I. hungarica</i> , <i>I. pseudacorus</i>
Isomangiferin	0.55	–	–	dark yellow	bright orange	green	<i>I. imbricat</i> , <i>I. sibirica</i> , <i>I. hungarica</i> , <i>I. pseudacorus</i>

Then coumarins (coumarin, umbelliferone, aesculetin, scopoletin, isoscoupoletin, herniarin, daphnoretin) were separated from rhizomes of *Iris pseudacorus* and the results were presented in the paper<sup>28</sup>.

**Xanthones (C6 – C1 – C6).** In UV-light on chromatograms the substances were identified by distinctive dark-yellow light emission that is specific for benzo- $\gamma$ -pyrones<sup>8, 26</sup>. These bands were exposed to

ammonia vapours and got bright-orange, after processing chromogenic reagent these bands got green which confirms they are xanthones. In all four species of irises, mangiferin (R<sub>f</sub> 0.52; system 2; R<sub>f</sub> 0.49; system 10) and isomangiferin (R<sub>f</sub> 0.38; system 2) were identified.

**Tannins ((C6 – C3 – C6)<sub>n</sub>).** Extracts from rhizomes of irises were dark-purple that is specific for the group of hydrolysable tannins.

Table 4. The quantitative content of the main groups of BAS in the rhizomes of *Iris*es

Groups of BAS	Species of irises			
	<i>I. pseudacorus</i>	<i>I. imbricata</i>	<i>I. sibirica</i>	<i>I. hungarica</i>
Flavonoids (SP) <sup>1</sup>	2.10 ± 0.04	3.70 ± 0.07	1.29 ± 0.02	3.70 ± 0.07
Isoflavonoids (SP) <sup>2</sup>	2.18 ± 0.02	1.16 ± 0.02 <sup>3</sup>	2.08 ± 0.04	1.16 ± 0.02
Phenolic compounds (SP) <sup>4</sup>	2.82 ± 0.03	2.75 ± 0.05	2.60 ± 1.75	2.77 ± 0.05
Hydroxycinnamic acids (SP) <sup>5</sup>	0.68 ± 0.02	4.16 ± 0.07	1.72 ± 0.03	6.54 ± 0.12
Tannins (TM)	6.21 ± 0.03	10.90 ± 0.19	13.95 ± 0.25	10.59 ± 0.18
Sum of $\gamma$ -pyrones (SP) <sup>6</sup>	0.21 ± 0.01	0.52 ± 0.01	0.01 ± 0.05	0.85 ± 0.04

SP – spectrophotometrical method, TM – titrimetric method

<sup>1</sup>is in terms of rutin

<sup>2</sup>is in terms of onosid

<sup>3</sup>is in terms of ononin

<sup>4</sup>is in terms of gallic acid

<sup>5</sup>is in terms of chlorogenic acid

<sup>6</sup>is in terms of mangiferin

### Isolation and determination of phenolic compounds

To separate and identify the individual phenolic compounds, division was used of their sum on a column with silica gel or preparative chromatographic separation on paper in several systems of solvents with subsequent separated crystallization of the separated compounds. Extracts of the raw materials of irises were received by maceration with 50% alcohol, evaporated under vacuum to 700 ml. The received residual was sequentially processed in a separating funnel with chloroform, ethyl acetate and *n*-butanol. The received fractions were evaporated and applied on silica gel, as the eluents used chloroform, alcohol-chloroform mixture in various combinations and ethyl alcohol. The process of elution was monitored by paper and thin-layer chromatography in systems 1, 2, 3, 10 and 11. Identical fractions were combined, solvent was distilled, and substances were recrystallized from ethanol. The following substances were separated: caffeic, ferulic, chlorogenic, neochlorogenic acids; kaempferol, hispidulin; genistein, daidzein; mangiferin and isomangiferin.

### Quantitative analysis

The results of quantitative determination of biologically active substances in rhizomes of irises are presented in Table 4.

The highest content of flavonoids and hydroxycinnamic acids was pointed out for *Iris hungarica* and *Iris imbricat* more 3.5% and 4%, respectively. While isoflavonoids were contained more in beardless irises (more than 2%), all irises contain the sum of  $\gamma$ -pyrones to 1%. The high content of tannins in the raw material (from 6 to 14 %) explains the historical use of rhizomes of irises for the tanning of leather<sup>13)</sup>.

### Conclusion

In the rhizomes of *I. pseudacorus*, *I. imbricat* *I. sibiric* , *I. hungarica* from 7 to 21 substances of phenolic nature were revealed by paper and thin-layer chromatography, among them there were phenolcarboxylic acids, coumarins, flavonoids, isoflavonoids. In the rhizomes 15 substances were identified: gallic, coumaric, cinnamic, chlorogenic, neochlorogenic, ferulic, caffeic acids;

kaempferol, quercetin, hispidulin, daidzein, genistein, formononetin, mangiferin and isomangiferin. Later on, these substances were separated from the rhizomes of irises. The quantitative content of phenolic compounds, flavonoids, isoflavonoids, hydroxycinnamic acids,  $\gamma$ -pyrones, and tannins was determined. Detailed phytochemical analysis of irises was carried out for the first time. With the exception of kaempferol and hispidulin in the rhizomes of *Iris pseudacorus*, all substances have been identified in the investigated species of irises for the first time.

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